

Current Insights on Reaction Kinetics and Mechanisms of PFAS Destruction during Hydrothermal Alkaline Treatment (HALT)

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Hydrothermal Alkaline Treatment (HALT)

- PFAS destruction process in **subcritical water**
 - Temperature: 200 to 350 °C
 - Pressures: 10 to 25 MPa
 - Reaction times: 1.5 to 30 min
 - NaOH concentrations: 0.1 to 5 M-NaOH
- Highly effective at breaking down long-chain, short-chain and ultra-short chain PFAS
- Final products of reaction are inert salts (e.g., sodium fluoride, sodium carbonate, sodium sulfate)

Technical Development To-Date

Feedstock	Testing Done	Total DRE (%)
Dilute AFFF	Pilot-Scale	>99%
Concentrated AFFF	Lab-Scale	>99.99%
Fire training pond water	Pilot-Scale	>99.9%
Landfill leachate	Lab-Scale	>99%
Sorbent regen. brine	Pilot-Scale	>99.9%
Foam fractionate	Pilot-Scale	>99.9%



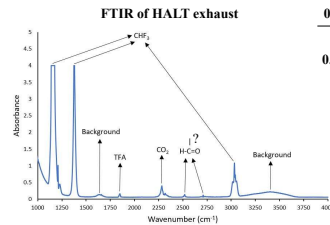
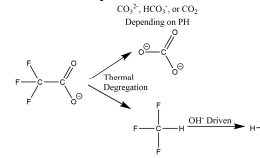
- Pilot Series system has been operated >200 hours for feasibility testing
- Steed Series (10 - 20 gallon per hour) units are being fabricated for field demonstrations in 2023 and 2024
- Efficacy demonstrated on several key feedstocks at-scale, with >99% destruction efficiency reliably achieved for concentrated feedstocks

Abstract

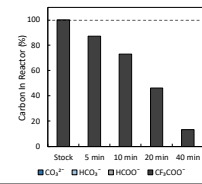
Hydrothermal alkaline treatment (HALT) is an emerging PFAS destruction process which facilitates **rapid and complete destruction** of all PFAS compounds. Current research has shown that a combination of thermal decomposition, hydrolysis, and OH⁻-driven reactions occur to convert PFAS species to final products of fluoride, carbonate, and sulfonate salts

Mechanism of Perfluorocarboxylic Acid Destruction in HALT

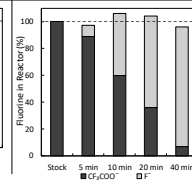
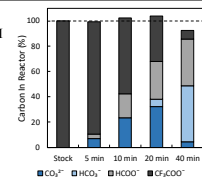
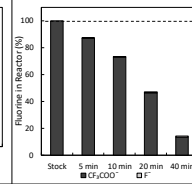
Proposed Reaction Mechanism



Carbon Balance



Fluoride Balance



Operating conditions for all bar charts: T = 200 °C, 0.1 M TFA

- TFA is destroyed under HALT conditions with and without NaOH
- NaOH is required to achieve **full mineralization**, breaking down the fluoroform (CF₃H) intermediate
- Initial reaction step is a thermal step to cleave off the -COOH head group, which occurs around 150 to 250 °C

Federal Funding & Support



HALT of Foam Fractionate in Pilot System



Analyte	Foamate (ppb)	HALT Effluent (ppb)	Destruction %
PFOS	36,400	10.9	99.97%
PFHpS	522	0.199	99.96%
PFHxS	6,260	3.34	99.95%
PFPeS	1,160	1.09	99.91%
PFBS	552	1.58	99.71%
PFOA	484	0.047	99.99%
PFHpA	334	ND	>99.993%
PFHxA	2,540	ND	>99.999%
PFPeA	196	ND	>99.98%
6:2 FTS	8,800	ND	>99.999%
Total PFAS	58,280	17.16	99.97%

- AFFF-impacted water from a fire training pond treated with bench-scale foam fractionator. Foamate treated with HALT
- >99.9% destruction of total PFAS
- >99% destruction of all individual PFAS
- No detectable intermediate products

Conclusions

- HALT destruction mechanisms are becoming better understood, with recent testing indicating thermal mechanisms are active for PFCAs
- >99% destruction efficiency can be reliably achieved for all PFAS in pilot-scale HALT reactor