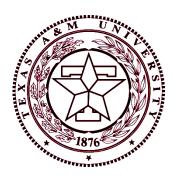
Biodegradation of Fluorotelomer-Based PFAS by Soil Cultures Enriched with Various Carbon Sources



Kung-Hui (Bella) Chu, P. E.

Professor, Zachry Department of Civil & Environmental Engineering Texas A&M University, College Station, TX <u>kchu@civil.tamu.edu</u>; https://chulab.engr.tamu.edu

Presentation to 2023 Battelle Bioremediation Symposium at Austin, Texas May 9, 2023





Project Team

Dr. Kung-Hui (Bella) Chu

Texas A&M University (TAMU)

Dr. Jennifer Field

Oregon State University (OSU)

Dr. Timothy J. Strathmann

Colorado School of Mines (CSM)

Graduate students and postdoc/senior scientists

- Shih-Hung (Jason) Yang, Jinha Kim, Jason Shih TAMU
- Inga Van Meter, Mitchell Kim, Ivan Titaley, Scott Leonard OSU
- Ori Soker -CSM

Supported by

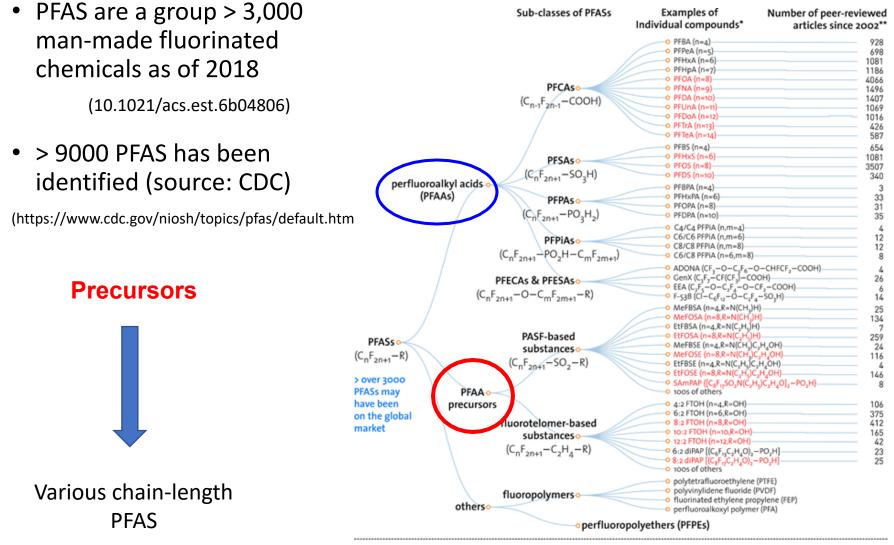








Per- and Polyfluoroalkyl Substances (PFAS)



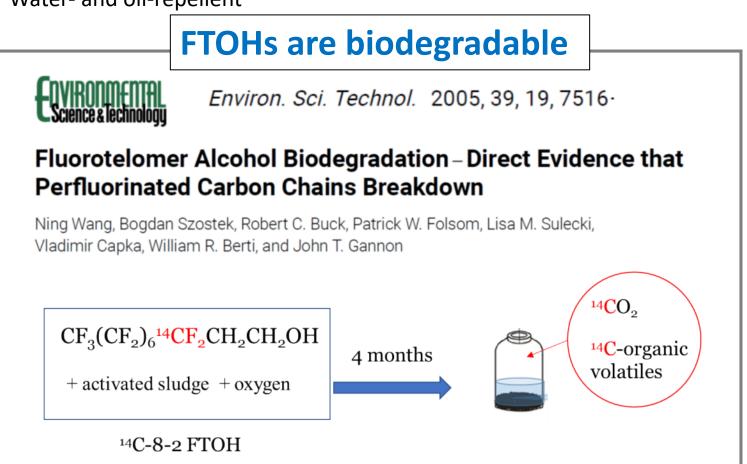
 PFASs in RED are those that have been restricted under national/regional/global regulatory or voluntary frameworks, with or without specific exemptions (for details, see OECD (2015), Risk reduction approaches for PFASs. http://oe.cd/1AN).

** The numbers of articles (related to all aspects of research) were retrieved from SciFinder® on Nov. 1, 2016.

FTOH-based PFAS

- Man-made fluorinated chemicals
- Unique property
 - Thermally stable and chemically inert
 - High surface activity
 - Water- and oil-repellent

Per – fully fluorinated Poly – partially fluorinated

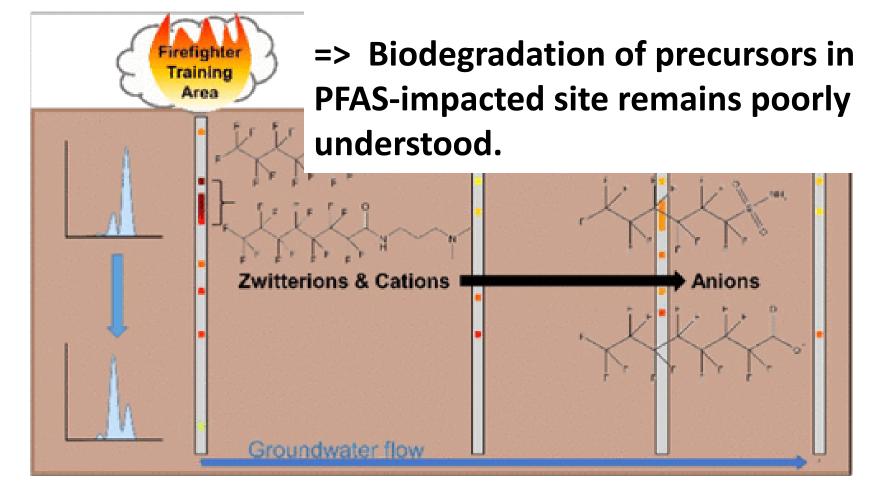




pubs.acs.org/est

Spatial Trends of Anionic, Zwitterioni AFFF-Impacted Site

Anastasia Nickerson, Alix E. Rodowa, David T. Adamsor John J. Kornuc, and Christopher P. Higgins* **52%** of the PFAS transported downgradient was associated with **polyfluorinated precursors**



Question

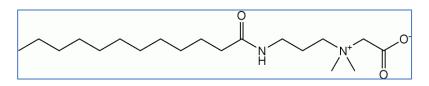
• Will biostimulation or bioaugmentation promote rapid bioconversion of the precursors?

Soil Enrichment Cultures With Different Carbon Sources

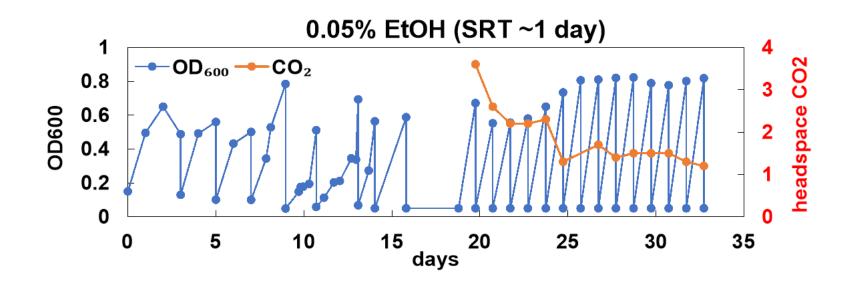
- C2 Background soil from a former air force base
- Carbon sources.
 - ✓ ethanol (EtHO, 0.05%),
 - ✓ 1-propanol (1-PrOH, 5mM),
 - ✓ 1-butanol (1-BuOH, 5mM),
 - ✓ hexane (0.05%),
 - ✓ octane (0.05%)
 - ✓ phenol (0.05%)
 - ✓ cocamidopropyl betaine (CPB) (0.05%).
- Ammonia mineral salts (AMS)
- Room Temp, 150 rpm

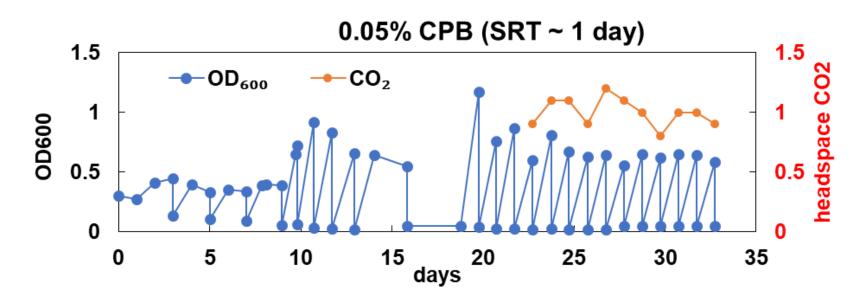
СРВ

- A zwitterionic surfactant
- Present in AFFF formula : 1-5% w/w.

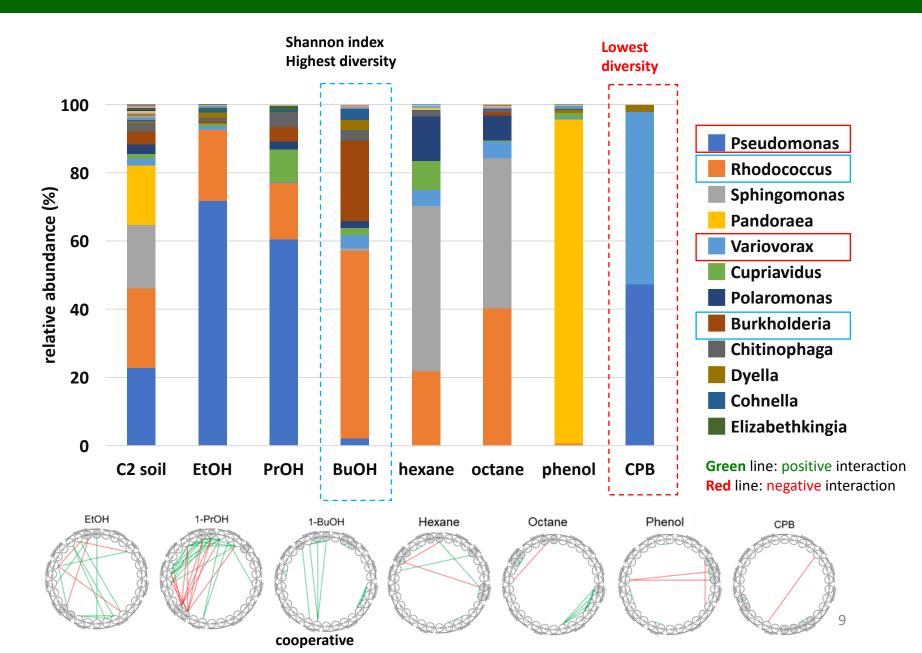


Enrichments in Sequential Batch Reactors

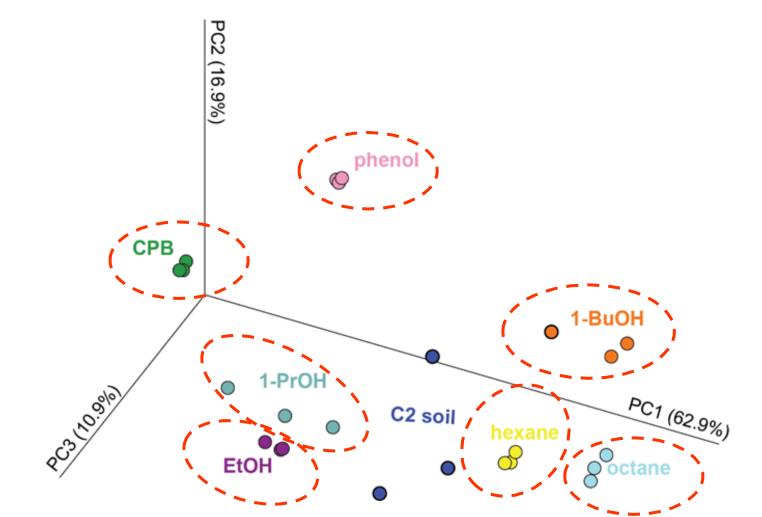




Microbial Communities of the Enrichments

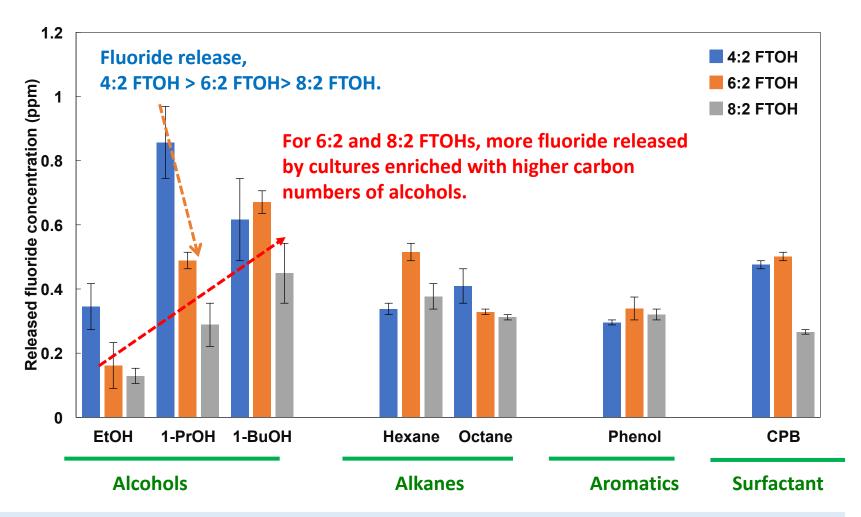


Carbon Sources are the Drivers of Different Microbial Communities of the Enrichments



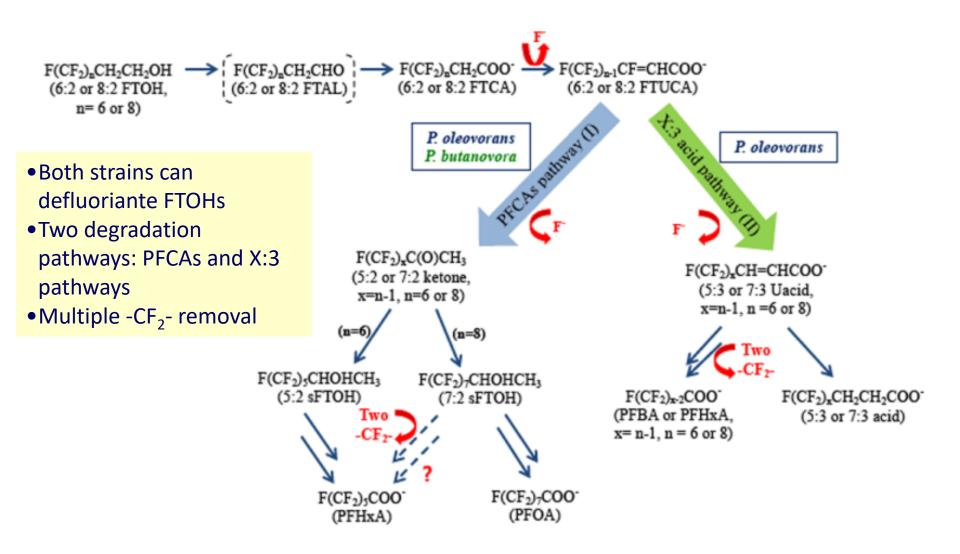
Principle coordinate analysis (PCoA) plot on the weighted UniFrac distance matrix referring beta-diversity of C2 soil and enrichment sample microbial communities from respective carbon sources with sequence depth of 33,000.

Biodegradation of FTOHs by the Enrichments

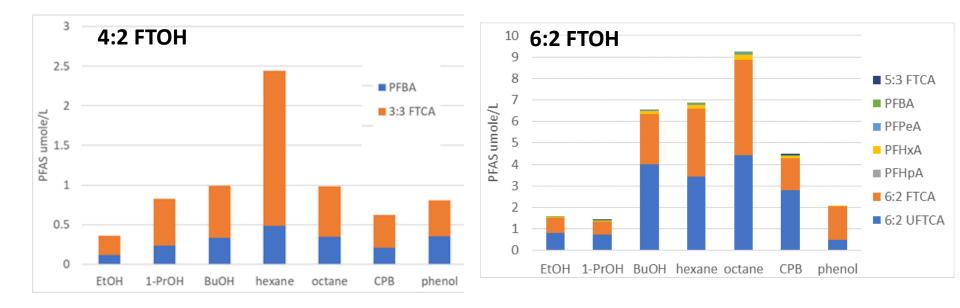


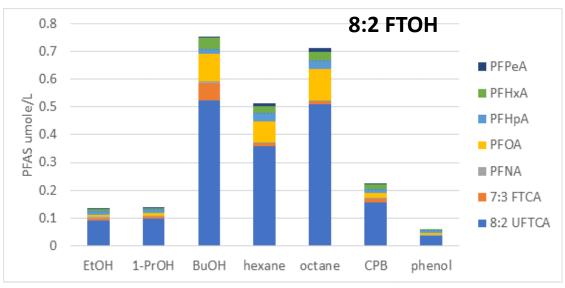
- All the enrichment cultures were able to degrade FTOHs.
- No chain-length dependent defluorination of FTOHs was observed for alkane- phenol- and CPB-enriched cultures.

Two Pathways for FTOH Biodegradation

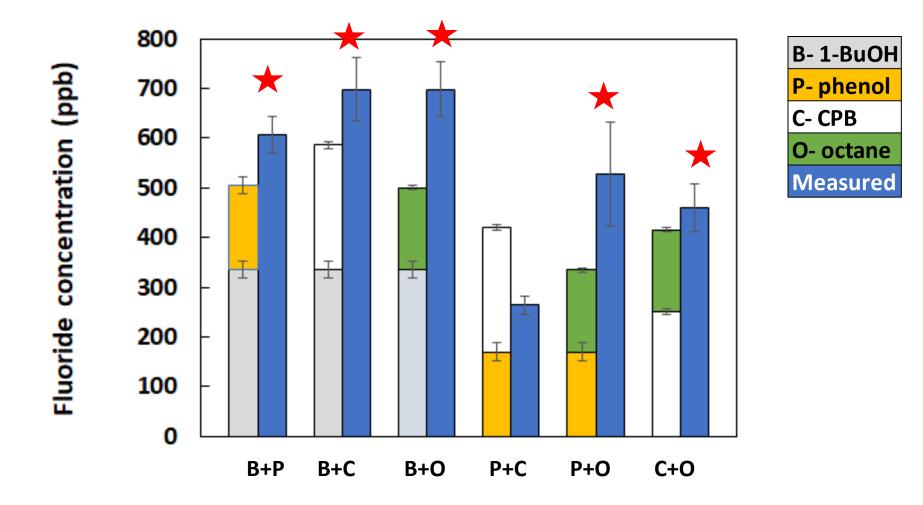


Biodegradation Metabolites of FTOHs





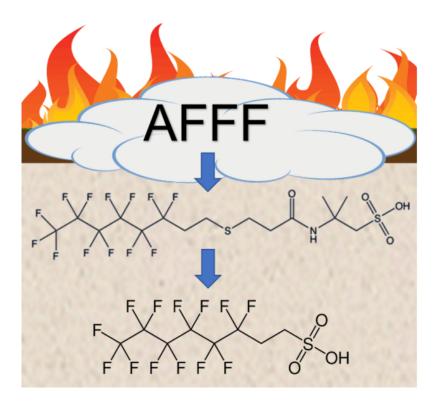
Enhanced Defluorination of 6:2 FTOH by Dual Enrichments



Higher 5:3 FTCA found in combinations with CPB-enriched culture

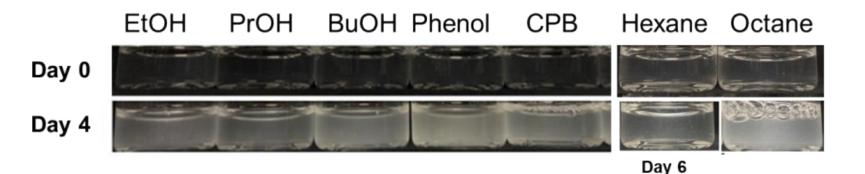
6:2 FTSA in AFFF-impacted Soil

- 6:2 Fluorotelomer sulfonates (FtS, FTS, or FTSA)
 - as ingredients in aqueous filmforming foams (AFFFs)
 - as transformation metabolites from precursors in AFFFs
- Known Precursors to fluorotelomer sulfonates
 - fluorotelomer thioether amido sulfonate (FtTAoS)
 - fluorotelomer sulfonamide betaines (such as 6:2 FtSaB)

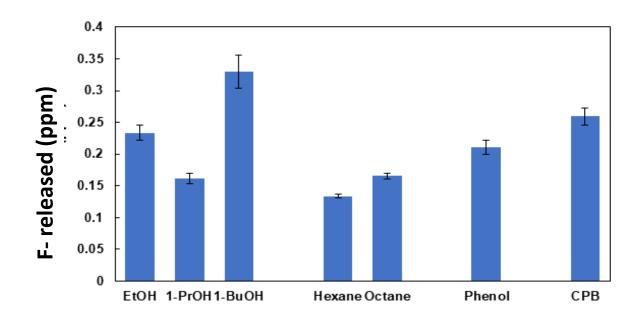


- 6:2 FTSA found in high levels in AFFF-impacted soils and groundwater
 - AFFF-impacted soil: 612-2,101 μg/kg
- 6:2 FTSA is also detected in
 - River, groundwater and wastewater: 1.6-37.9 ng/L
 - Landfill leachate: 582 ng/L

Enrichments Used 6:2 FTSA as S Source

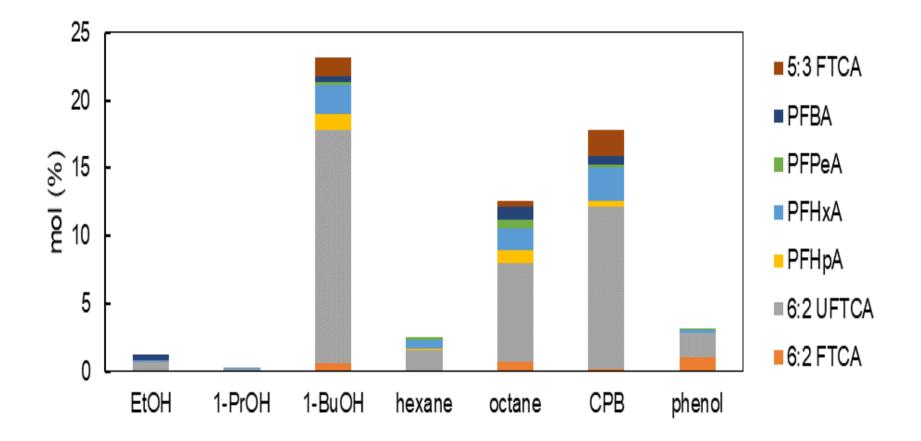


• F⁻ released in spent growth medium (initial 6:2 FtS =20 ppm)



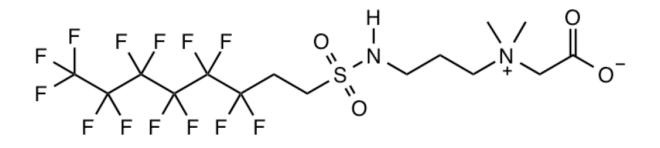
Metabolites of 6:2 FTSA when 6:2 FTSA was used as S source by the Enrichments

Metabolites (initial 6:2 FtS concentration = 4 ppm)



Can Enrichments Degrade 6:2 FtSaB?

- 6:2 fluorotelomer sulfonamide alkylbetaine (6:2 FtSaB or 6:2 FTAB)
 - a novel perfluorooctane sulfonate alternative
 - induced developmental toxicity in zebrafish embryos
 - 6:2 FTSA can be produced as its metabolite

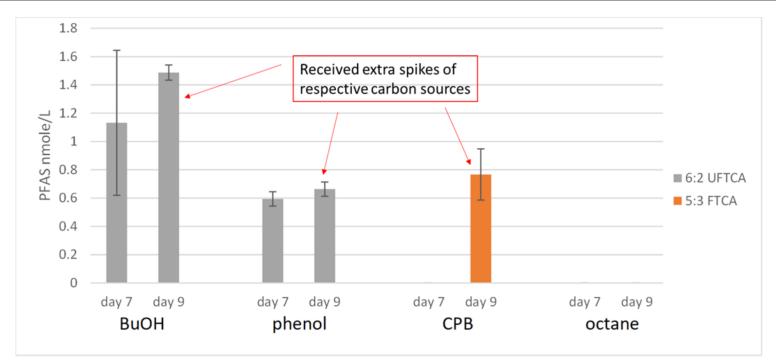


 Under sulfur-rich conditions, no fluoride release was detected after two weeks when 6:2 FtSaB was spiked into the enrichment cultures.

Can enrichments use 6:2 FtSaB for Growth?

Exp. Setup:

- Butanol-, octane-, phenol- and CPB-enriched cultures + 4 mg/L of 6:2 FtSaB + respective carbon source in S-free medium on day 0.
- Two additional carbon source spikes on day 3, and day 7 for butanol-, phenol- and CPB-enrichment cultures.
- One additional carbon source spike on day 5 for octane-enrichment culture.



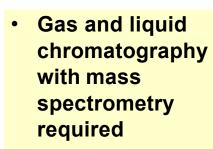
Capstone 1157

Capstone 1157: a complex commercial reference material comprised of:

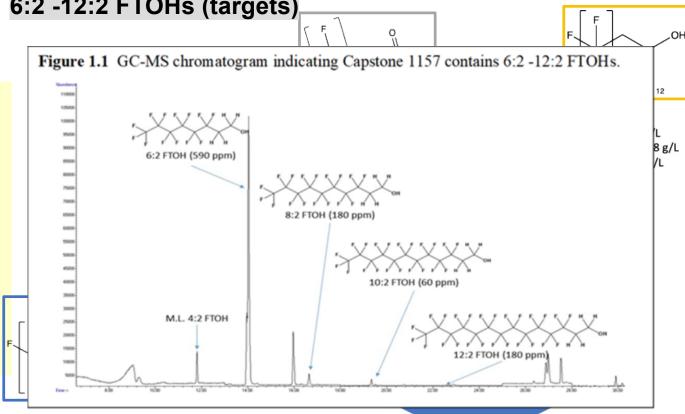
- Non-volatile PFAS:
- 6:2 12:2 FtSaBs (1 target)
- 6:2 8:2 FtSaAms (suspects)
- 4:2 10:2 FtS (targets)



• Volatile PFAS: 6:2 -12:2 FTOHs (targets)



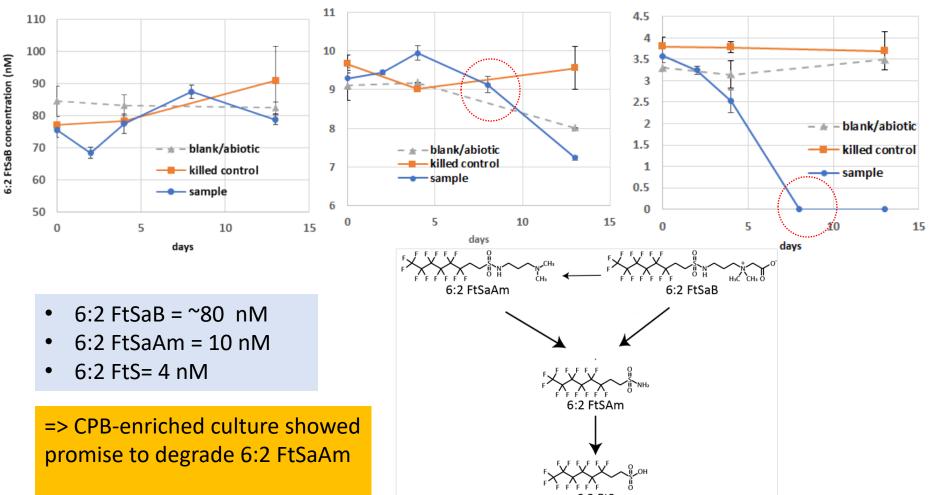
 Combination of suspect & target data analysis



Biodegradation of Capstone 1157

- Two enrichment cultures: CBP- and butanol-enriched cultures
- S-limited conditions 6:2 FtSaB

^{6:2} FtS



^{6:2} FtSaAm

Take Home Message

- Soil enrichments with different carbon sources (alcohols, alkanes, phenol, and CPB)
 - Different carbon sources shaped different microbial community structure., with butanol-enriched culture showed the most diversity and CPB-enriched culture showed the least.
 - All enrichments can defluroinate FTOHs to different extent.
 - Chain-length dependent defluorination of FTOHs was observed for alcohol-enrichment cultures. Also, extent of defluroination of FTOHs : 4:2 > 6:2 > 8:2 FTOH.
 - enhanced F- release from 6:2 FTOH degradation by dual enrichements.

Take Home Message

- Soil enrichments with different carbon sources (alcohols, alkanes, phenol, and CPB) (conti)
 - All enrichment cultures can used 6:2 FtS as sole Ssource and defluorinated them under S-limited conditions
 - Butanol- and CPB-enriched cultures can use 6:2 FtSaB as sole S- source and defluorinated them under Slimited conditions
- Capstone contains both volatile and non-volatile PFAS precursors
 - n:2 FtSaB (n= 6, 8, 12) > n:2 FtSaAm (n= 6, 8) > 6:2 FtS > 6:2 FTOH.

